## UNIVERSITY OF MEDICAL SCIENCES, ONDO

## **DEPARTMENT OF PHYSIOLOGY**

# PHS 212: BLOOD AND BODY FLUID PHYSIOLOGY LECTURER: MR A.O. AKINOLA

### **OBJECTIVES**

- Introduction
- Definition of body fluids and body fluid compartments
- Physiological variation of body fluid volumes
- List the ionic composition of different body compartments
- Regulation of body fluid volumes
- Techniques for quantifying various body fluid volumes

#### Introduction

Water is the major constituent of all body fluid compartments.

➢ In an adult human, average of 70-kilogram

Total body water is about 60 per cent of the body weight, or about 42 liters.

#### PHYSIOLOGICAL VARIATION OF BODY FLUID VOLUME

Most of the variation in body fluid between individuals is as a result of variation in amount of body fat or adipose tissue (fat is only about 10% water)

- ➢ Infant: 73-80%
- ➤ Male adult: 60%
- ➢ Female adult: 40-50%
- ➢ Effects of obesity
- ➢ Old age 45%



#### Normal Daily Fluid Input and Output

Inputs	Outputs
Ingestion	≻Gut (Faeces 100ml)
Fluid (1.25 Liters)	≻Urine (1.5 Liters)
Food (1 Liter)	➢Breathing/Skin (900ml)
≻Metabolism (350 ml)	Sweating (100ml)
≻Total (2.6 Liters)	≻Total (2.6 Liters)

#### Abnormal Fluid Input and Output

Inputs	Outputs
► Injection/Infusion	≻Gut
≻Hyperdipsia	<ul> <li>Vomiting</li> </ul>
	<ul> <li>Diarhoea</li> </ul>
	≻Skin- Burns
	≻Sweating-
	Excessive
	➤Urine- Diabetes insipidus
	≻Haemorrhage



PERCENTAGE OF WAT	ER IN TISSUES
Average 70 kg person	total body weight
▶ 42 litres total H2O	60%
28 liters Intracellular fluid (ICF)	40%
14 liters Extracellular fluid (ECF)	20%

ECF is divided into <sup>3</sup>/<sub>4</sub> ISF and <sup>1</sup>/<sub>4</sub> plasma water

- ➤ 10.5 liters Interstitial fluid (ISF)
  15%
- ➤ 3.5 liters Plasma water
  5%
- All the transcellular fluids together constitute about 1 to 2 liters.

#### **IONIC COMPOSITION OF BODY FLUIDS (mEq/l)**

Substance	ICF	ISF	Plasma
Cations			
Na <sup>+</sup>	14	140	142
K <sup>+</sup>	150	5	5
Ca <sup>2+</sup>	2	5	5
$Mg^{2+}$	30	2	2
Anions			
Cl-	10	113	113
HCO <sub>3</sub> -	10	27	27
HPO <sub>4</sub> -/H2PO <sub>4</sub> -	113	2	2
SO <sub>4</sub> <sup>2-</sup>	20	1	1
Protein	74	2	16

#### **REGULATION OF BODY FLUID VOLUME**

The hypothalamus achieve this by two mechanisms:





#### THIRST MECHANISM

- > Thirst centre in the hypothalamus contains osmoreceptors
- > When ECFV decreases then osmolarity increases
- Some of the stimulated and this activate the thirst centre
- ➤ Thirst is initiated
- $\succ$  The person feels thirsty and drink water
- ➢ ECF volume increases and ECF osmolarity decreases

#### ADH MECHANISM

- > When ECFV decreases, ECF osmolarity increases
- Supraoptic nucleus in the hypothalamus is stimulated
- > ADH is released
- > ADH causes retention of water by facultative reabsorption
- ECFV increases and osmolarity decreases
- > On the contrary, when ECFV increases
- ➢ No ADH is released, decrease water reabsorption

#### **MEASUREMENT OF BODY FLUID VOLUME**

- Can be done both directly (indicator dilution method) and indirectly
- > By injecting substances that will stay in one compartment
- Then calculate the volume of fluid in which the test substance is distributed (volume of distribution of injected material)
- ➢ Volume of distribution
  - = <u>amt. of substance injected- amt. excreted/metabolized</u> concentration of the substance in the sample

Example: 150mg of sucrose is injected into a 70kg man. The plasma sucrose level after mixing is 0.01mg/ml, and 10mg has been excreted or metabolized during the mixing period. Calculate the volume of distribution of sucrose

#### CHARACTERISTICS OF MARKER SUBSTANCE/INDICATOR

- Must be non-toxic
- > Must mix evenly throughout the compartment being measured
- Must have no effect of its own on the distribution of water or other substances in the body
- It must be unchanged by the body during mixing period, or the amount changed must be known
- $\succ$  It must be easy to measure

#### MEASUREMENT OF TOTAL BODY WATER

- ≻Using indicator dilution method
- ► Indicators include:

- ➢ Radioactive water (tritium, <sup>3</sup>H<sub>2</sub>O) or heavy water (deuterium, <sup>2</sup>H<sub>2</sub>O) and antipyrine
- > They mix with TBW within few hours of injection

#### MEASUREMENT OF ECF VOLUME

- ➢ Using indicator dilution method
- Indicators include:
- Radioactive sodium, radioactive chloride, thiosulphate ion, inulin, mannitol and sucrose
- The substances mix almost completely throughout ECFV within 30-60mins

#### MEASUREMENT OF PLASMA VOLUME

≻Using indicator dilution method

- Indicators include: Evan's blue (T-1824) and radioactive iodine(<sup>131</sup>I)
- > The substance binds with plasma proteins strongly
- ≻Diffuses into ISF in small quantity or not at all

INDICATORS USED FOR MEASURING PLASMA VOLUME, ECF VOLUME AND TOTAL BODY H<sub>2</sub>O

Compartment Criterion

Indicators

- Plasma Substance should not cross capillaries
   Plasma Substance should not cross or adioiodinated fibrinogen;
   radioiodinated albumin
- •ECF volume Substance should Isotonic solutions of sucrose, cross capillaries inulin, mannitol, NaCl but not cross cell membranes

•Total body H<sub>2</sub>O (TBW) SubstanceHeavy  $H_2O$ , tritiated  $H_2O$ ,distributes evenlyaminopyrine, antipyrinein ICF & ECF-

Blood volume /Markers used Obtained from plasma volume and hematocrit ➤ Total blood volume = <u>Plasma volume</u> 1- Hematocrit

Example: If the plasma volume is 4 liters and the hematocrit is 0.45, total blood volume is ?

#### Take this problem:

100 mg of sucrose is injected into a 70 kg man. The plasma sucrose level after mixing is 0.01 mg/ml. If 5 mg has been metabolized during this period, then, what is the ECF volume?

9.5 L

14 L

17.5 L

10 L

#### COMPARTMENTS WITH NO COMPARTMENT-SPECIFIC SUBSTANCE

Determine by subtraction:

- ≻How would you measure ICF volume?
- Cannot be measured; it is calculated (estimated)..
- > ICF volume = Total body H<sub>2</sub>O ECF volume

 Interstitial volume:
 ➢ Can not be measured directly
 ➢ Interstitial Fluid Volume (ISFV): ISFV = ECFV - PV

# Please emulate the biblical Bereans (Acts 17:11)

# **END OF LECTURE**